



Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/gmcl19>

Transient Fluorescent Characteristics of GFP/Viologen/Cytochrome-c Hetero-LB Films

Jeong-Woo Choi ^a, Yun-Suk Nam ^a, Dongho Kim ^b,
Won Hong Lee ^a & Masamichi Fujihira ^c

^a Dept. of Chem. Eng., Sogang Univ., C.P.O. BOX 1142, Seoul, 100-611, Korea

^b National Creative Research Initiatives Center for Ultrafast Optical Properties Characterization, Korea Research Institute of Standards and Science, P.O. BOX 102, Yusong, Taejeon, 305-600, Korea

^c Dept. of Biomol. Eng., Tokyo Institute of Technology, Yokohama, 227, Japan

Version of record first published: 24 Sep 2006

To cite this article: Jeong-Woo Choi, Yun-Suk Nam, Dongho Kim, Won Hong Lee & Masamichi Fujihira (2001): Transient Fluorescent Characteristics of GFP/Viologen/Cytochrome-c Hetero-LB Films, Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals, 371:1, 383-386

To link to this article: <http://dx.doi.org/10.1080/10587250108024765>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.tandfonline.com/page/terms-and-conditions>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Transient Fluorescent Characteristics of GFP/Viologen/Cytochrome-*c* Hetero-LB Films

JEONG-WOO CHOI^a, YUN-SUK NAM^a, DONGHO KIM^b,
WON HONG LEE^a and MASAMICHI FUJIHIRA^c

^a*Dept. of Chem. Eng., Sogang Univ., C.P.O. BOX 1142, Seoul 100-611, Korea,*

^b*National Creative Research Initiatives Center for Ultrafast Optical Properties Characterization, Korea Research Institute of Standards and Science, P.O. BOX 102, Yusong, Taejeon 305-600, Korea and* ^c*Dept. of Biomol. Eng., Tokyo Institute of Technology, Yokohama 227, Japan*

The charge transfer characteristics of molecular hetero-LB films composed of green fluorescence protein (GFP), viologen, and cytochrome-*c* were investigated by transient fluorescence measurement. In this work, transient fluorescence decay of GFP homo film and GFP/viologen/cytochrome-*c* hetero films were measured to investigate the charge separation rate and transport mechanism by using time-correlated single photon counting method (TCSPC).

Keywords: Green fluorescent protein (GFP); Cytochrome *c*; Time resolved fluorescent; Electron transfer; TCSPC

INTRODUCTION

In the biological photosynthesis, photoelectric conversion and long-range electron transfer occur not only efficiently but also unidirectionally through the functional groups of biomolecules. Electron sensitizer/electron acceptor structured hetero films have the similar structure with the photoinduced

electron transfer system of the bacterial photosynthetic reaction center^[1]. The green fluorescent protein (GFP) is the final light emitting protein in the bioluminescent jellyfish *Aequorea victoria*. GFP absorbs blue light and emits green light (510nm). Since GFP shows very highly efficient quantum yield, approximately 80%, it is very reasonable approach to use GFP as an electron sensitizer in elucidation of the electron transfer mechanism of natural biological photosynthesis.

In this work, the hetero-Langmuir-Blodgett (LB) films consisting of GFP/viologen/cytochrome-*c* was constructed. The transient fluorescence decay profile of hetero-LB films was measured to investigate the charge separation rate and transport mechanism.

EXPERIMENTAL DETAILS

Viologen (Electron relay; R) and cytochrome-*c* (Electron acceptor; A) films were deposited onto the quartz plate by LB method with a circular type Langmuir trough (Model 2022, Nima Tech.,UK). The viologen/cytochrome-*c* hetero-LB films were soaked into the GFP (Electron sensitizer; S) solution (1 μ M, pH8.0) for 1hr, and then the GFP molecules were electrostatically adsorbed onto the surface of hetero-LB films.

Transient fluorescence decay profiles were recorded by using TCSPC technique with a femtosecond Ti:Sapphire laser pulse excitation. The fluorescence lifetimes of S homo, S/R hetero, and S/R/A hetero films were determined from the response profiles at 510nm.

RESULTS AND DISCUSSION

Fig.1 shows the energy diagram of S/R/A structured hetero-films. By light illumination, the electrons of S molecules are excited from their ground

state to excited state (S^*). The photo-excited electrons return to their ground state and then green fluorescence is emitted at 510nm.

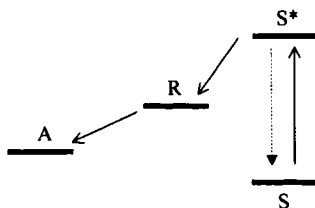


FIGURE 1. The energy diagram of S/R/A structured hetero-film

However, the electron acceptor is exposed to the excited sensitizer, some of photo-excited electrons of S^* can be separated ($S^+/R^-/A$) and transferred ($S^+/R/A^-$) to A molecule via their redox potential difference. Thus the photoinduced one-way electron transfer can be generated. By measuring the transient fluorescence, the charge separation rate of S molecule to A molecule could be calculated and the photoinduced one-way electron transfer could be verified.

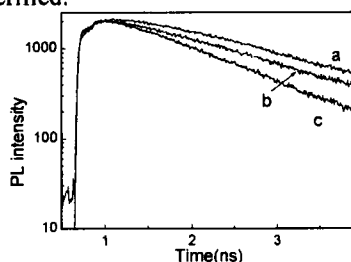


FIGURE 2. Transient fluorescence decay profiles: a, S homo-film; b, S/R hetero-film; c, S/R/A hetero-film

Transient fluorescence of each film was obtained at 510nm. The charge separation rate (k_a) of excited S to R/A can be written as $k_a = 1/\tau_t - k_s = 1/\tau_t - 1/\tau_s$. Where k_a is the electron transport rate from excited S to R/A, τ_t and τ_s are the fluorescence lifetime of S in hetero and homo-films, respectively, and k_s is the

fluorescence decay rate of excited S in homo-films. Based on the above equation, the charge separation rate can be obtained from the fluorescence lifetime of S in homo and hetero-film. The transient fluorescence decay profiles were shown in Fig.2.

The fluorescence lifetime of S homo-films was c.a 1.54ns. The time constant and charge transfer rate of each film were shown in Table 1. The fluorescence decay profile of S homo-film was well described as single exponential component, whereas the decay profiles of S/R and S/R/A hetero-film were well described as double exponential components.

TABLE 1. Time constants and charge transfer rates of each film

Films structure	Component 1	Component 2
S homo	1.54ns	-
S/R hetero	0.17ns	1.38ns
k_a (s ⁻¹)	5.25×10^9	-
S/R/A hetero	0.105ns	1.274ns
k_a (s ⁻¹)	8.87×10^9	-

The charge separation rate of S/R/A hetero-films was faster than that of S/R hetero and S homo-films. In this result, the one-way electron transport from S to A via R was verified, and this result can be extended to develop bioelectronic devices, such as a photodiode and memory device.

Acknowledgements

This work was supported by grants from Korea Science and Engineering Foundation (KOSEF : 98-0502-08-01-3) and National Creative Research Initiatives of Ministry of Science and Technology.

References

[1.] K.S. Cho, J.W. Choi, W.H. Lee, N.W. Song, and D. Kim, Molecular Crystals & Liquid Crystals, **327**, 275 (1999)